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 $2 \text{ mol C}_3\text{H}_6\text{O}_2 \rightarrow 2 \text{ mol H}_2\text{O}$. How many moles of H_2O are produced when 50.75 g of O_2 react? $50.75 \text{ g O}_2 \times \frac{1 \text{ mol O}_2}{32 \text{ g}} \times 2 = 3.17 \text{ mol H}_2\text{O}$
 $31.9988 \text{ g O}_2 \rightarrow 7 \text{ mol O}_2$. How many moles of O_2 are requir...

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