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[EXAMVIEW - QUIZ 3--HEAT AND THERMO PRACTICE.TST](#)

Name: Class: Date: ID: ACh 9 (Heat) & Ch 10 (Thermo) Quiz PRACTICE
 Multiple ChoiceIdentify the choice that best completes the statement or answers the question.
 1. What happens to the internal energy of an ideal gas when it is heated from 0C to 4C? a. It increases. b. It decreases. c. It remains constant. d. It is impossible to determine.
 2. Which of the following is proportional to the kinetic ene...

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[CSE543/FALL 2006 - QUIZ](#)

Thursday, November 16, 2006 Professor Trent JaegerPlease read the instructions and questions carefully. You will be graded for clarity and correctness. You have 25 minutes to complete this quiz, so focus on those questions whose subject matter you know well. Write legibly and check your answers before handing it in. Short Answer - some will be one or two words no more than 3 sentences
 1. (3pts) Why ...

File name: cse543-f06-quiz.pdf

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Chemistry 31 Quiz 3 10 minutes March 20Name Lab Section NumberH2C2O4 Ka1 = 5.62 x 10², Ka2 = 5.42 x 10⁵; Kw = 1.00 x 10⁻¹⁴. (5 pts) Calculate the pH of the following (Ignore activity and assume no secondary reactions besides acid base in water.):
 0.015 M Na2C2O4 [C2O4²⁻] = 0.015 M See Chapter 6 problems 46, 48, and 6-I
 1.00 x 10⁻¹⁴ C2O4²⁻ + H2O ? HC2O4⁻ + OH⁻ Kb1 = 5 = 1.845 x 10¹⁰ 5.42 x 10⁻¹⁰ [HC2O4⁻] [OH⁻]

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